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Vapor pressure - Wikipedia

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In a chemical reaction, chemical equilibrium is the state in which both reactants and products are present in concentrations which have no further tendency to change with time, so that there is no observable change in the properties of the system. Usually, this state results when the forward reaction proceeds at the same rate as the reverse reaction.

Chemical equilibrium - Wikipedia

2 Addition to double and triple bonds " Mechanism Hydration " Hydroboration " Hydroxylation " epoxidation. Elimination reactions E1, E2, E1cB Mechanism " Orientation effects in elimination reactions " stereo chemistry of elimination reactions - dehydration of alcohols " dehydro halogenation " cope

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